Review Article Radionuclide generators: the prospect of availing PET radiotracers to meet current clinical needs and future research demands

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Abstract: Targeted molecular imaging with positron emission tomography (PET) constitutes a successful technique for detecting and diagnosing disease conditions promptly and accurately, and for effectively prognosticating outcomes and treating patients with a tailored and more individualized intervention. In order to expand the success of PET in nuclear medicine, it is important to assure access to radiotracers of desired quantities and qualities. In this context, the benefit of accessing PET radiotracers through a radionuclide generator (RNG) cannot be overstated, as generators offer the potential of enriching the PET radiotracer arsenal at the medical centers both with and without onsite cyclotrons. While RNG technology to avail PET tracers is in its infancy, their use is expected to revitalize current PET practices and seems poised to broaden the palette of PET in nuclear medicine in the foreseeable future. In this review, we discuss the principles of RNGs, assess major parent/daughter pairs of interest for PET, RNGs currently in use in clinical PET, and identify the potentially useful RNGs which have made substantial progress or are likely to be used in daily clinical practices in the near future. Availability of the parent radionuclides required for PET RNGs is an important criterion and hence their production will also be reviewed. This overview outlines a critical assessment of RNGs to avail PET tracers, the contemporary status of RNGs, and key challenges and apertures to the near future.

Keywords: Bifunctional chelator (BFC), coronary artery disease (CAD), parent/daughter radionuclide, positron emission tomography (PET), radionuclide generator (RNG), radiopharmaceuticals

Introduction

The role of PET to enable in vivo visualization of physiological processes on the molecular level in real time and quantify them by measuring regional concentration of the radiotracer for diagnosing disease, monitoring disease progression, and tracking therapeutic response, hardly needs to be reiterated [1-12]. This momentous molecular imaging paradigm, straddling the disciplines of molecular biology and medical imaging technology, has not only brought a perpetual shift in healthcare practice but also heralds a significant leap forward in treatment outcomes [13-15]. Growth in the field of PET has been phenomenal and will continue in the foreseeable future. The expanding role of PET in nuclear medicine (NM) procedures has led to an exponential growth of the research and development (R&D) efforts around the world. The driving force behind the success has been largely due to the rapid ascent of the positron emission tomographycomputed tomography (PET/CT) system which offers anatomic (CT) as well as metabolic (PET) information, in addition to providing data for attenuation correction [16, 17]. While the utility of PET/CT lives at the interface between many scientific disciplines, cost effective availability of PET tracers of required quality and quantity is a key determinant that underpins survival, strength, and success of the modality. Currently, ¹⁸F, ¹⁵O, ¹³N and ⁶⁸Ga are the most commonly used PET tracers, among which ¹⁸F has dominated significantly and has been regarded as the workhorse of PET in clinics [3, 6, 12, 18-22]. Reflecting on the last decade of development on PET, one can clearly see the impact of ¹⁸F

which was blended into a number of formulations to improve current imaging practices and overall performance. Although the use of these tracers in clinical PET constitutes successful advancement in the field and heralds a new era of molecular imaging, the requirement of an onsite cyclotron to produce these tracers, due to their short half-life, has emerged as the major impediment that continue to thwart efforts for their widespread use in daily NM routines. In this context, the prospect of accessing PET tracers through an RNG seems to be a promising proposition as it enables PET examinations at remote hospitals and at the same time offers the prospect of enriching the PET radiotracer arsenal at the medical centers both with and without onsite cyclotrons. The tremendous prospects associated with the use of RNGs, along with the challenge of providing PET tracers of requisite quality for a variety of NM diagnostic procedures, have led to a considerable amount of fascinating research and innovative strategies, the flow of which shows no sign of diminishing. With to the aim of obtaining PET tracers in an acceptable chemical form amenable for the formulation radiopharmaceuticals through the use of an RNG, essentially every conceivable strategy has been exploited.

Rapid and burgeoning research interests in the use of RNGs in PET have been the driving force to provide a detailed review on this subject, in an attempt to stimulate interest in this exiting field. In order to improve the utility of RNGs in PET, it is of utmost importance to nurture emerging RNG technologies in an appropriate manner to facilitate their transition from laboratory research to the clinical settings. The aim of this article is thus to provide an overview of RNGs to avail PET radiotracers which are currently in use in clinical practice, or which have made substantial progress or are likely to be materialized in the foreseeable future. In the following sections, we provide a detailed coverage on the different types of RNGs available to obtain PET tracers, their contemporary status, and their aperture to the near future. Given the multi-disciplinary field, speculative options reported mainly of academic interest are not included and the authors apologize for possible oversights of important contributions. This review is intended to serve as a resource for the researchers and to offer an impetus for further development, as well as to become familiar with the expectations, capabilities, constraints, and gratification involved in the development of RNGs for today and tomorrow.

Radionuclide generator

A RNG is a self-contained system housing an equilibrium mixture of a parent/daughter radionuclide pair and designed to provide the daughter radionuclide formed by the decay of the parent radionuclide [23-29]. The parent/daughter nuclear relationships offer the possibility of separating the daughter radionuclide at repeated time intervals [30]. The inherent determinant for the success of RNGs resides with the selection of parent/daughter pairs and appropriate radiochemical separation strategies, which are based on a number of considerations [31]. Overviews of the principle, criteria for selection of parent/daughter pairs, and growth and equilibrium of the daughter radionuclide with parent radionuclide have been elaborately discussed in detail in recent reviews [26, 30-32]. The striking diffusion and the exciting perspective of RNGs in NM are mainly attributed to the following causes:

RNGs ensure onsite availability of PET tracers on demand and without reliance on an onsite accelerator, thereby, representing a cost-effective proposition for the onsite formulation of radiopharmaceuticals.

Offer the prospect of availing PET tracers in nocarrier-added (NCA) form.

Provides the flexibility to perform multiple studies, due to ready availability of multiple doses of PET tracer on-demand.

Constitute the only on-site option of availing certain short-lived radionuclides (⁸²Rb) which cannot be shipped as with commercial sources.

Several requirements need to be fulfilled for effective separation of the daughter radionuclide, and in general the process should be fast, reproducible and provide the daughter radionuclide of required radionuclidic, radiochemical and chemical purity with a high radiochemical yield [23-27, 29-32]. A wide range of separation procedures, each with different characteristics, are currently being used or can potentially be used for RNG technology. Overviews of

Concreter	Half life		β⁺ branch		Application	
Generator	Parent	Daughter	(%)	$E_{\beta+}$ / wev	Application	
⁶⁸ Ge/ ⁶⁸ Ga	270.8 d	1.14 h	89.0	0.74	Labelling Perfusion	
⁸² Sr/ ⁸² Rb	25.6 d	1.27 min	95.0	1.41	Perfusion	
⁴⁴ Ti/ ⁴⁴ Sc	60.3 y	3.927 h	94.0	0.597	Labelling	
⁶² Zn∕ ⁶² Cu	9.26 h	9.74 min	97.0	1.28	Labelling, perfusion	
¹¹⁰ Sn/ ^{110m} In	4.1 h	1.15 h	62.0	0.623	Labelling	
⁷² Se/ ⁷² As	8.4 d	1.083 d	88.0	1.02	Labelling	
$^{140}Nd/^{140}Pr$	3.37 d	3.39 min	51.0	0.544	Perfusion	
¹¹⁸ Te/ ¹¹⁸ Sb	6.00 d	3.6 min	74.0	0.882	Perfusion	
¹²² Xe/ ¹²² I	20.1 h	3.6 min	77.0	1.09	Perfusion	
¹²⁸ Ba/ ¹²⁸ Cs	2.43 d	3.62 min	69.0	0.869	Perfusion	
¹³⁴ Ce/ ¹³⁴ La	3.16 d	6.4 min	63.0	0.756	Perfusion	
⁵² Fe/ ^{52m} Mn	8.28 d	21.1 min	97.0	1.13	Perfusion	

 Table 1. Key examples of RNGs to provide positron emitting radionuclides with potential for PET imaging

radiochemical separation processes with respect to RNGs are elaborated in recent reviews [26, 27, 31, 32]. In-growth of the daughter radionuclide is continuous and depends on its half-life as well as the frequency of separation. For practical considerations, RNGs are eluted at periodic intervals, depending on the daughter activity requirements.

Table 1 presents examples of RNG systems capable of providing positron-emitting daughter nuclides relevant for quantitative PET. Generator produced, short half-life PET radionuclides (on the order of minutes) do not allow for radiochemical synthesis and are mainly used for characterizing the faster kinetics of smaller tracer molecules. The longer-lived PET radionuclides are better suited to studying the slower kinetics of labeled peptides, antibodies and cells.

Potentially useful PET RNGs for biomedical applications

The interest in RNGs will vary according to the scenarios considered. The following section provides an overview of the potential RNGs that could be used to avail PET tracers for clinical and research needs.

⁶⁸Ge/⁶⁸Ga generator

In recent years, the ⁶⁸Ge/⁶⁸Ga generator has evoked excitement among radiopharmaceutical researchers and captured the imagination of NM physicians, thanks to recent progress in

PET instrumentation, hybrid imaging modalities, and advances in molecular and cellular biology [33-40]. The parent ${}^{68}\text{Ge}$ (t_{1/2} = 270 d) radionuclide decays by EC to 68 Ga (t₁₆ = 68 min) which subsequently decays to stable 68Zn via positron emission and electron capture (branching ratios: $\beta^+ = 89\%$, EC = 11%). The nuclear transitions are also accompanied by low intensity gamma emission $[E_g = 1077 \text{ keV}]$ (3%)]. Decay characteristics of the 68Ge/68Ga generator are depicted in Figure 1.

The use of ${}^{68}\text{Ge}/{}^{68}\text{Ga}$ generators in PET is very attractive for the following reasons:

The 270-day half-life of ⁶⁸Ge ensures the ability to use the generator for extended periods, potentially up to 1 year or longer.

The decay characteristics of the short-lived daughter $^{68}Ga~(t_{_{1/2}}$ = 68 min, 89% β^+ and 11% EC, E $_{_{\beta^+,~max}}$ = 1.92 MeV) are convenient for PET imaging.

With a physical half-life of 68 min, ⁶⁸Ga also matches the biological half-life of many peptides used for imaging, due to their rapid diffusion, localization at the target and fast blood clearance.

The ability of ⁶⁸Ga³⁺ to form stable complexes with many ligands containing oxygen and nitrogen as donor atoms offers the prospect for complexation with a wide variety of chelators, as well as some macromolecules having significant clinical potential.

Commercial ⁶⁸Ge/⁶⁸Ga generators have a longer history than many people would believe. Although ⁶⁸Ge/⁶⁸Ga generators have been investigated since 1960s, the chemical forms of generator derived ⁶⁸Ga and the unacceptable level of ⁶⁸Ge breakthrough emerged as the major obstacles that continued to thwart efforts for the development of gallium radiochemistry and the practical application. The first ⁶⁸Ge/⁶⁸Ga RNG based on liquid-liquid extraction technology, using acetyl-acetone in



cyclohexane, was described in 1960 [41]. Due to the inherent of limitations of liquid-liquid extraction, and the requirement of three to four separation steps in order to obtain the requisite purity amenable for the preparation of ⁶⁸Ga compounds for radiopharmaceuticals applications, subsequent developmental efforts were directed toward the use of column chromatography techniques. The first ⁶⁸Ge/⁶⁸Ga generator based on a column chromatography technique [42] consisted of an Al₂O₃ matrix from which ⁶⁸Ga was eluted as a strong ⁶⁸Ga-EDTA complex using a 5-mM EDTA solution at pH 7.0. While the use of alumina based ⁶⁸Ge/⁶⁸Ga generators is prolific in providing 68Ga-EDTA complex to measure the increased blood flow of brain tumors, it requires decomposition of the EDTA for the preparation of other 68Ga-radiopharmaceuticals because of the high stability of the complex (log K = 21.7). This extra step is not only cumbersome but also results in the loss of useful 68Ga radioactivity. In an attempt to mitigate this shortcoming, Kopecky et al. [43] eluted the column with 0.2 M HCl with a 48% elution yield. Malyshev et al. [44] used ZrO₂, SnO₂, and TiO₂ matrices and eluted ⁶⁸Ga with either HCI, HNO₃, or CH₃COOH as eluents. Of the three-column matrices used, ZrO, was found to be the best, in which 68Ga was eluted with 0.1 M HCl with 35% elution yield. Subsequently Arino et al. [45] used a polyantimonic acid column matrix and eluted 68Ga with 2% sodium oxalate. The poor elution yield of ⁶⁸Ga, the toxicity of the eluent, and the presence of oxalic acid emerged as the major impediments that prevented clinical use without further chemical manipulation. A generator based on the tin dioxide/HCl couple has also been tried by Loc'h et al. [46]. In all these metal oxide

matrix-based systems, the release of column matrix due to limited solubility in the eluent constituted a major limitation which ruled out all possibility of clinical use. With the aim of circumventing this limitation, use of organic adsorbents and elution of ⁶⁸Ga with dilute HCI or HF has also been tried, but met with limited success [47, 48]. McElvany et al., in 1984 [49], evaluated three ⁶⁸Ge/ ⁶⁸Ga generators based on different metal oxide matrices

over a period of 1 year and compared their performances with respect to ⁶⁸Ga elution profiles and yields, parent 68Ge breakthrough levels, amounts of column matrix contaminants present in the generator eluate, and the ease of ⁶⁸Ga-radiopharmaceutical production. The ionic 68Ga/68Ga generator utilizing a tin dioxide column eluted with 1 M HCl was found to be the most suitable for preparation of ⁶⁸Ga-labeled compounds for use in conjunction with PET. In another embodiment, α -Fe₂O₃, activated carbon, and graphite were tried [50], and it was found that $\alpha\text{-Fe}_{2}\text{O}_{3}\text{,}$ having a ^{68}Ga elution yield of 50-70% with a HCl solution of pH 2.0, was the most suitable. In spite of the great publicity received by 68Ga imaging, its impact in NM started to fade in the late 1970s due to the commercial non-availability of 68Ge/68Ga generators adaptable to the syntheses of a wide range of ⁶⁸Ga radiopharmaceuticals, in comparison to the parallel and rapid developments of several new classes of radiopharmaceuticals based on ^{99m}Tc and ¹⁸F. While the utility of ⁶⁸Ge/⁶⁸Ga generators for clinical ⁶⁸Ga imaging would be in hibernation for decades, the foundation needed to build to the next level has been well established and separation procedures developed for the isolation of ⁶⁸Ga from ⁶⁸Ge with HCl solutions proved to be fertile ground for the realization of a generator for direct application in a clinical context.

The first commercial ${}^{68}\text{Ge}/{}^{68}\text{Ga}$ generator based on a modified TiO₂ solid phase support is the product of years of hard work by scientists of Cyclotron Ltd, Obninsk, Russian Federation [51]. Each generator contains ${}^{68}\text{Ge}$ activities of up to 3.7 GBq from which "ionic" ${}^{68}\text{Ga}^{3+}$ can be availed in 0.1 N HCl solution with ~80% initial

Years	Generator type	Remarks
1960-1970	liquid-liquid extraction	Tedious time consuming separation; not amenable for hospital radiopharmacy.
1970-2000	Column chromatography system based on ${\rm Al_2O_3}$ and EDTA as eluent	This system served as a convenient and economical source of ⁶⁸ Ga-EDTA; clinical use was limited to measure increased blood flow of brain tumors, in particular.
After 2000	Column chromatography system based on TiO_2 , ZrO_2 , CeO_2 , SnO_2 or an organic resin	⁶⁸ Ga is availed in an ionic form, with elution yields 70% to 80%; ⁶⁸ Ge breakthrough still in the range, 0.01-0.001%.

Table 2. Types of ⁶⁸Ge/⁶⁸Ga generators

Table 3. Commercial ⁶⁸Ge/⁶⁸Ga generators

Manufactures	Туре	Year	Column matrix	Eluent	Elution yield	⁶⁸ Ge breakthrough
Eckert & Ziegler AG, Berlin, Germany	Obninsk	1996	TiO ₂	0.1 M HCI	60-75%	< 0.01%
	IGG100	2008	TiO ₂	0.1 M HCI	70-75%	< 0.001%
iThemba, South Africa		2008	SnO ₂	0.6 M HCI	80%	< 0.002%
ITG Garching, Germany		2010	Silica based organic matrix	0.05 M HCI	> 80%	< 0.005%



Figure 2. Commercially available ⁶⁸Ge/⁶⁸Ga generators according to column matrix, manufacturers and the year of introduction to the market.

 68 Ga elution yields and with a 68 Ge break-through of about 1 × 10⁻³%.

Commercial success of this generator, together with the advancement in imaging techniques, drew new players to enter the market. ⁶⁸Ge/⁶⁸Ga generator development during the last few decades is briefly summarized in **Table 2**. Currently, four different ⁶⁸Ge/⁶⁸Ga generators are commercially available and only a few of them (Eckert & Ziegler; iThemba Labs) hold license for good manufacturing practice (GMP) production. It is pertinent to mention that a draft monograph, "Gallium (⁶⁸Ga) chloride solution for radiolabeling", commensurate with the regulatory requirements for quality and patient safety, is available [52]. Characteristics of major commercial ⁶⁸Ge/⁶⁸Ga generators available today for use in NM centers are tabulated in **Table 3** and shown in **Figure 2**.

Despite remarkable advancements, the low radioactive concentration, high [H⁺] burden, ⁶⁸Ge breakthrough, and the presence of poten-



 Table 4. 68 Ga labeled radiopharmaceuticals substituting 99mTc

Diagnostics modality	Established ^{99m} Tc labeled agents	Potential ⁶⁸ Ga labeled agents
Peptide receptors	^{9m} Tc-HYNIC-peptide	⁶⁸ Ga-DOTA-peptide
Bone metastases	^{9m} Tc-MDP	⁶⁸ Ga-phosphonates
Renal function	^{9m} Tc-DTPA/MAG3/DMSA	⁶⁸ Ga-EDTA
Cardiac function	^{9m} Tc-RBC/MIBI	68Ga-BAPEN
Lung function	^{9m} Tc-MAA	⁶⁸ Ga-MAA
Hepatobiliary	^{9m} Tc-IDA	⁶⁸ Ga-IDA
Infection	^{9m} Tc-WBC	68Ga-citrate
Brain imaging (perfusion)	^{9m} Tc-ECD	⁶⁸ Ga-ECD

cation and/or concentration procedures based on (1) anion-exchange chromatography, (2) cation-exchange chromatography and (3) solvent extraction have been developed to obtain ⁶⁸Ga of acceptable radionuclidic purity and radioactive concentration [53-59]. Among them, the chromatographic procedure reported by Zhernosekov et al. [54], based on selective tr-

tial metal ions such as Al, Fe, Cu, Zn, Ti or Sn from generator eluate have emerged as the major deterrents in the path of direct labeling of ⁶⁸Ga to make radiotracers from generator availed 68Ga. Although the concentrations of these metallic impurities are low (~ ppm level), their concentration can still be much higher than that of NCA 68Ga3+, impeding efforts for the preparation of ⁶⁸Ga-labeled radiopharmaceuticals. Most of the commercially available ⁶⁸Ge/⁶⁸Ga generator systems demonstrate deteriorating performance in terms of increased ⁶⁸Ge breakthrough and reduced ⁶⁸Ga elution yields on repeated elutions over prolonged periods of time [53, 54]. In order to circumvent such limitations, a variety of post-elution purifi-

apping of ⁶⁸Ga eluate on a cation-exchanger, followed by elution in a small volume of acetone-HCl mixture, was a major step and has reached the stage of commercial exploitation. The trace level of acetone present in the final reaction mixture could be averted by making use of the column-based radiochemical procedure reported by Loktionova et al. [57]. Even though there is no technical impediment to adapt this efficient and robust procedure, the requirement of an automated module is viewed as a necessary one not only to minimize the decay loss of ⁶⁸Ga but also to reduce the radiation exposure. Nevertheless, the scope of using this technique is restricted to TiO, and organicresin based generators due to the explicit need

Diagnostics modality	Established ¹¹ C/ ¹⁸ F labeled agents	Potential 68Ga labeled agents
Angiogenesis	¹⁸ F-galacto-RGD	68Ga-DOTA-RGD, 68Ga-VEGF
General cancer imaging	18FDG	68Ga-CXCR4 biomarker, 68Ga-uPAR biomarker, 68Ga-SCN-NOTA-BZA
Нурохіа	¹⁸ F-nitroimidazloes (FAZA, FMISO, FETNIM)	68Ga-DOTA-imidazoles
Proliferation	¹⁸ FLT	68Ga-D03A-thymidine
Glioma	¹⁸ FET, ¹¹ C-methionine	68Ga-glutamine, 68Ga-DO3A-alanine, 68Ga-DO2A-tyrosine
Prostate cancer	¹⁸ FDG, ¹¹ C-acetate, ¹⁸ F-choline, ¹¹ C-choline	68Ga-DOTA-PSMA

 Table 5. 68Ga labeled radiopharmaceuticals substituting 18F, 11C labeled compounds



of primary ⁶⁸Ga eluate in HCl concentration \leq 0.1 N for effective sorption in the cationexchange column. In the case of SnO₂ based generators, where ⁶⁸Ga is eluted in 0.6-1 N HCl, the aforementioned post-elution processing can be effectively adapted after appropriate dilution. Commercial availability of ⁶⁸Ge/⁶⁸Ga generators yielding the ionic form of ⁶⁸Ga, along with a post-elution processing system, has led to the exploration of a broad spectrum of ⁶⁸Ga-labeled products for use as radiopharmaceuticals for PET imaging of cancer [60].

The availability of ⁶⁸Ge/⁶⁸Ga generators providing 68Ga3+ of requisite quality amenable for direct radiolabeling is a much more desirable option. In this regard, the potential of the BARCdeveloped ⁶⁸Ge/⁶⁸Ga generator, based on the nanoceria-polyacrylonitrile (CeO₂-PAN) composite [61, 62] sorbent, is capable of yielding ⁶⁸Ga of requisite quality (free from metal ion impurities such as Al, Fe, Cu, Zn, Ti or Sn ions and very low 68Ge breakthrough) and without the need for any post-elution processing, and is poised to significantly expand the scope of ⁶⁸Ga products formulation in radiopharmacy. Based on a similar strategy, a nano zirconia based ⁶⁸Ge/⁶⁸Ga generator has also been developed, which demonstrates comparable performance

[63]. A schematic diagram of the 'BARC' ⁶⁸Ge/⁶⁸Ga generator is shown in **Figure 3**. Over the years, the 'BARC' ⁶⁸Ge/ ⁶⁸Ga generator has been effectively developed and refined for use in the preparation of ⁶⁸Ga tracers, both for research and limited clinical use. A comprehensive quality assurance system is necessary to ensure that the quality of ⁶⁸Ga availed from the generator meets the standards for clinical use.

⁶⁸Ga is exclusively in the oxidation state III in aqueous solution as well as at physiological pHs. In light of the perceived need to preclude the formation of insoluble Ga(OH), and soluble Ga(OH)⁴⁻, synthesis of ⁶⁸Ga-labeled radiopharmaceuticals is carried out in the presence of weakly coordinating ligands such as citrate, acetate, or oxalate, which dramatically reduce the kinetics of complex formation [64]. Gallium (III) is classified as hard Lewis acidic, and therefore binds to hard Lewis base donor atoms such as nitrogen and oxygen, generally to form six coordinate bonds in close to octahedral geometry. Several promising ⁶⁸Ga tracers ccomprising of small molecules, large biomolecules, and particles, targeting biological activities such as proliferation, angiogenesis, and apoptosis are currently under preclinical and clinical investigation [37, 60, 65-73]. Given the very broad field and long history of ⁶⁸Ga tracers. agents that are being tested in preclinical and clinical trials, which promise an exciting future in clinical PET, are included in this review.

While ⁶⁸Ga has made significant inroads into the field of clinical PET and undergone phenomenal expansion and growth, one of the most rapidly expanding areas is the development of ⁶⁸Ga labeled peptide-based agents for targeted imaging, of tumors in particular [60]. The use of



68Ga labeled small tumor-affine peptides targeting somatostatin receptors (SSTR) has not only changed the diagnostic approach to neuroendocrine tumors (NETs) such as pituitary adenoma, pancreatic islet cell tumor, carcinoid, pheochromocytoma, paraganglioma, medullary thyroid cancer, and small cell lung carcinoma, but also has unlocked the potential of these agents for molecular imaging applications [37, 38, 72, 74-78]. Among the SSTR agonists, ⁶⁸Ga-labelled DOTA-conjugated somatostatin analogues have emerged as the breakthrough vector molecules owing to in vivo stability, favorable pharmacokinetics, and high and specific receptor-mediated tumor uptake [60]. They have proven to be increasingly useful due to several technical and biological advantages. such as fast clearance, rapid tissue penetration, and low antigenicity. The possibility of using 68Ga-labelled DOTA-conjugated somatostatin analogues for diagnosis is enticing, because the same compound labeled with 90Y or 177Lu can be used for therapy as a theranostic paradigm. This not only provides the prospect for

better planning of therapy but also offers the opportunity to evaluate the therapeutic outcome, as in personalized medicine. This will allow for dosimetry before therapy to ensure the optimum balance between risk and benefit by enabling prediction and avoidance of potential radiotoxicity. The Society of Nuclear Medicine and Molecular Imaging (SNMMI) announced on November 18, 2013 that, 68Ga-DOTATOC has been officially designated an orphan drug by the U.S. Food and Drug Administration for the management of neuroendocrine tumors (NET) [79]. This is considered as a potential step in right direction that will contribute not only to potentially faster regulatory approval but to streamlined clinical trials.

The currently available data for potential use of other 68Ga-labeled peptide analogs as PET tracers are mainly preclinical and limited human studies have been carried out. In this context, 68Ga-labeled DOTA-4-amino-1-carboxymethylpiperidine-D-Phe-GIn-Trp-Ala-Val-Gly-His-Sta-Leu-NH2 peptide (BAY86-7548), having a high affinity for the bombesin receptor subtype II for the detection of primary and metastatic prostate carcinoma, merits attention [80, 81]. Integrin $\alpha_{\alpha}\beta_{\alpha}$ is an important member of receptor family and expressed preferentially on regenerative vascular endothelial cells and some tumor cells. Preliminary clinical studies indicate that integrin is an effective target for detecting intra-prostatic prostate cancer [82]. ⁶⁸Ga-labeled α_.β₂ integrin-targeting ⁶⁸Ga-c[Lys-(NOTA)-Arg-Gly-Asp-D-Phe] is one of the radiolabeled arginine-glycin-aspartic acid (RGD) peptides for which initial clinical data are available. A biodistribution and radiation dosimetry study containing 10 patients having lung cancer or lymphoma showed that the excretion route with the highest activity was found to be the renal pathway. A clinical trial on ⁶⁸Ga-BNOTA-PRGD [68Ga-p-SCN-Bn-NOTA-PEG3-RGD] through intravenous injection of a single dose of nearly 111 MBg (\leq 40 µg BNOTA-PRGD_a), is in progress to investigate post-myocardial infarction and post-stroke repair. Preliminary clinical studies indicate that 68Ga-PRGD2 uptake was found at or around the ischemic region in both MI and stroke patients, and significantly correlated with the disease phase and severity [83].

Urea-based low-molecular-weight peptidomimetic inhibitors of the prostate-specific membrane antigen (PSMA), a which is a cell surface



Figure 6. Schematic representation of ⁸²Rb infusion system that can control concentration of ⁸²Rb activity administered to the patient, perform quality assessment and flush the activity from the patient line at the end of the infusion.

protein and is expressed at higher levels in prostate cancer compared to other tissues, constitute a promising target for specific imaging due to the protein's transmembrane location and internalization after ligand binding [84-86]. ⁶⁸Ga-Glu-NH-CO-NH-Lys-(Ahx) [⁶⁸Ga-PSMA-11] has emerged as an attractive agent currently used in clinical studies for the detection of recurrent prostate cancer and metastatic spread [86].

The enormous success of ⁶⁸Ga-labelled peptides served as a springboard to spur the development of other ⁶⁸Ga-labelled radiopharmaceuticals, which is primarily based on two main trends. First is the technetium "shake and shoot" concept, which primarily relied on the substitution of ⁶⁸Ga in lieu of ^{99m}Tc in the preparation of radiopharmaceuticals. Some of the ⁶⁸Ga labeled radiopharmaceuticals which can be substituted for ^{99m}Tc are shown in **Table 4**. A few examples of this type include the use of commercially available phosphonate kits for bone scintigraphy of lesions, MAA (macroaggregated human serum albumin) particulate for perfusion studies, ⁶⁸Ga-NOTA-MSA (mannosylated human serum albumin) for immune system imaging, and ⁶⁸Ga-BAPEN myocardial uptake as substitute for MIBI [87]. The second trend is the development of ⁶⁸Ga labeled radiopharmaceuticals in place of ¹⁸F and ¹¹C labeled compounds with an aim to achieve on-line cyclotron independence. These are elaborated in **Table 5**.

Several innovative radiopharmaceuticals labeled with ⁶⁸Ga are under active investigation and it is anticipated that quite a few novel agents labeled with this positron-emitting radionuclide will be available in the near future [88, 89]. With expanding areas of application and growing interest in the use of 68Ga labeled radiopharmaceuticals, it is simply a matter of time until imaging of bone, sentinel lymph node, and/or lung ventilation/perfusion will be carried out routinely by ⁶⁸Ga/PET-CT. Conspicuous harnessing of the 68Ga tracers in conjunction with their convergence with theranostics seems poised to bridge the quantitative diagnosis with subsequent therapeutic management. This bridging will benefit from therapeutic strategies aimed at the same or closely related processes



Figure 7. Photograph of the CardioGen-82 Infusion System along with various part of the system.

and may be of great value for supporting and strengthening the concept of personalized medicine. It is time to bring this innovative paradigm changing concept out of the shadows and into the light.

⁸²Sr/⁸²Rb generator

There is a great deal of interest on the use of ⁸²Sr/⁸²Rb generators to avail ⁸²Rb for clinical myocardial perfusion PET investigations. ⁸²Rb is the first of the generator-produced positron emitters that made its entry into clinical NM. ⁸²Rb, with a half-life of 76 s, decays by positron emission into stable ⁸²Kr, which is a noble element and is therefore non-reactive and safe for biological use. Decay characteristics of the ⁸²Sr/⁸²Rb generator are depicted in **Figure 4**.

Being a cationic analogue of potassium with similar chemical and biological properties, ⁸²Rb accumulates, as a function of blood flow, in cells of myocardium and other tissues in a manner similar to potassium. The myocardial uptake of ⁸²Rb reflects blood flow through the myocardium. Compared with normal myocardium, areas of ischemia or infarction exhibit low ⁸²Rb uptake because of diminished blood flow and/or viability, and this is useful for qualitative infarct imaging and for the detection of coronary artery stenosis and characterization of the severity. Synthesis of ⁸²Rb radiopharmaceuticals for diverse imaging applications is precluded not only by the relatively limited chemistry of the alkali metals, but also due to the short physical half-life of ⁸²Rb [90]. The exciting perspective of ⁸²Sr/⁸²Rb generators in clinical PET is attributed mainly due to the following:

The short half-life of ⁸²Rb (t_{y_2} = 76 s) enables rapid rest/stress paired studies within a very short time (30-45 min), allowing for rest and stress imaging under virtually identical conditions and decreasing the total time required to scan each patient. This is convenient for the patient and permits high-throughput imaging and efficient use of the technology.

The ultra-short half-life of ⁸²Rb presents limited radiation exposure to the patients (whole body dose of 5.5 mSv for 2.22 GBq) during investiga-



Figure 9. Post elution processing of ^{44}Sc obtained from a $^{44}\text{Ti}/^{44}\text{Sc}$ generator.

tion. The low-dose (10 MBq/kg) protocol allows simultaneous quantification of absolute flow, on 3D PET systems with adequate dynamic range to permit accurate measurement of the bolus first-pass activity, and measurement of ventricular function close to peak hyperemia with pharmacologic stress.

The suitably short physical half-life of ⁸²Rb not only allows administration of high doses and short imaging times, but also offers the prospect of performing repeated and sequential perfusion studies within 20-minute intervals. The 25.36 d half-life of parent radionuclide ⁸²Sr is convenient for shipping and provides useful radioactivity in a generator for at least a month of frequent elutions.

Use of an RNG provides the scope to perform PET investigations in medical institutes without the necessity to maintain expensive cyclotron facilities. ⁸²Rb has significant clinical potential and can be a considered as a low-cost alternative to short-lived, cyclotronproduced PET isotopes.

⁸²Rb is extracted from the plasma with high efficiency by myocardial cells via the Na⁺/ K⁺ ATPase pump [91]. While the myocardial extraction of ⁸²Rb is similar to that of ²⁰¹Tl, it is slightly less than ¹³NH₂. The uptake of ⁸²Rb is a function of blood flow, metabolism, and myocardial cell integrity. Cardiac PET using ⁸²Sr possesses several advantages over traditional SPECT, with direct patient centric benefits including the accurate, diagnosis of coronary artery disease in asymptomatic or symptomatic patients, assessment of coronary stenosis severity, myocardial infarct imaging, evaluation of myocardial viability, collateral function, and cardiomyopathy.

In 1989 the United States Food and Drug Administration (FDA) approved the use of ⁸²RbCl availed from a commercial ⁸²Sr/⁸²Rb generator system trade name CardioGen-82, which has been available from Bracco Diagnostics Inc. for some years. This generator is composed of a small chromatography column [~4 cm (I) × 0.5 cm (Φ)] containing hydrous SnO₂ housed in a lead shielding container. A schematic diagram of the ⁸²Sr/⁸²Rb generator system is depicted in **Figure 5**. When the column is flushed with a solution, such as 0.9% NaCl saline, the ⁸²Rb⁺ is displaced by Na⁺ and is eluted. Quality control



Figure 10. Schematic diagram of ⁶²Zn/⁶²Cu generator to avail ⁶²Cu for radiopharmaceuticals formulation.

of the generator eluate includes ⁸²Sr breakthrough measurements using a dose calibrator and pyrogen tests of the eluate.

As the ⁸²Rb supply is 90% replenished within 5 to 10 minutes of the previous elution, serial studies can be performed in rapid succession. maximizing patient throughput. There appeared to be enticing interest in the use of an automated infusion system to administer the ⁸²Rb eluted from the 82Sr/82Rb generator directly to patients, in light of the short half-life of 82Rb and the decreasing amount of available ⁸²Rb as the generator ages [92]. In the quest for an effective strategy to avail consistent levels of ⁸²Rb activity from the ⁸²Sr/⁸²Rb generator, Klein et al. have developed an automatic infusion system [93]. A schematic representation of such a conceptual ⁸²Rb infusion system that can control the concentration of ⁸²Rb activity administered to the patient, perform quality assessment, and flush the activity from the patient line at the end of the infusion is shown in **Figure 6**.

Automated 82Rb infusion systems capable of accurate measurement and delivery of adequate doses of ⁸²RbCl from a ⁸²Sr/⁸²Rb generator have been developed by Bracco Diagnostics, NJ, USA (CardioGen-82 Infusion System) and Jubilant DraxImage Inc, Canada (Ruby-Fill™ Infusion System). 82Rb is eluted from the generator by a computer-regulated elution pump and infused directly into patients using commercially available IV infusion system. Use of these systems substantially reduces the radiation dose, ensures optimum performance of the ⁸²Sr/⁸²Rb generator, and provides a log of the 82Rb activity infused into the patients. These developments are considered as successful steps forward in the promotion of widespread clinical use of 82Rb. A photograph of the CardioGen-82

Infusion System, along with various part of the system, is shown in **Figure 7**. Such ⁸²Rb infusion systems clearly outperform conventional infusion systems or semi-automated systems, and their usefulness has been convincingly proven. Commercial availability of automated ⁸²Rb infusion systems has a far-reaching effect on realizing the full promise and potential of ⁸²Rb in the diagnosis of coronary artery disease (CAD) and is poised to bring functional imaging to the forefront in cardiac PET imaging.

⁸²Rb permits clinical imaging with short protocols (20-30 min in total) and a high patient throughput, providing better image quality and overall sensitivity in the diagnosis of coronary artery disease (CAD) as compared to myocardial scintigraphy with ²⁰¹Tl or ^{99m}Tc-based radiotracers [94-96]. It also offers the potential for deriving absolute quantification of myocardial blood flow (MBF) [97]. New approaches to

Component	H ₂ PTSM kit	H ₂ ETS kit	H ₂ ATSM kit
Sodium acetate	4.3 mg	4.3 mg	4.3 mg
Acetic acid	3.1 mg	3.1 mg	3.1 mg
Sodium chloride	30 mg	30 mg	30 mg
Excipient	20 mg	20 mg	20 mg
Free ligand	2 µg	2 µg	0.4 µg
⁶² Cu labeled ligand (per 555 MBq (15 mCi) dose)	2.3 × 10 ⁻⁷ mg	2.2 × 10 ⁻⁷ mg	2.4 × 10 ⁻⁷ mg

Table 6. Components of H₂PTSM, H₂ETS and H₂TASM kits

Table 7	. Possible	routes f	or the	production	of	⁸² Sr
	. 1 0001010	100001		production	0.	0.

Reaction	Target material	Projectile energy (MeV)	Comment
⁸⁹ Y (p, spallation) ⁸² Sr	Yttrium oxide	60-240	Low radiopurity & yield
^{Nat} Mo (p, spallation) ⁸² Sr	Molybdenum metal	500-700	Low radiopurity, high cost
^{Nat} Rb (p, xn) ⁸² Sr	RbCl or Rb metal	40-90	Preferred
$^{\text{Nat}}$ Kr (α , pxn) 82 Sr	Kr gas	20-120	Low radiopurity, low yield, little availability
^{Nat} Kr (³ He, xn)	Kr gas	20-90	Low radiopurity, low yield, very little availability

kinetic modeling and software tools for analysis of clinical ⁸²Rb studies are being developed and evaluated, which could enable quantification of absolute myocardial blood flow and flow reserve.

Use of generator availed ⁸²Rb in PET myocardial perfusion imaging has become far more accessible in daily practices, offers clinical feasibility of positron-based cardiac imaging without a cyclotron [98], and is poised to take a major leap forward to an exciting new stage. While the use of ⁸²Rb availed from ⁸²Sr/⁸²Rb generators is a successful paradigm for the diagnosis of obstructive coronary artery disease (CAD) and has made considerable progress, the primary impediment for its widespread clinical use is the expense associated with ⁸²Sr production and the need for generator replacement at 3-5 week intervals. Conscientious utilization of the generator in high throughput will not only offset this cost but can foster sustainability. The improved accuracy of PET leads to total cost savings in CAD management in clinical practice by getting rid of unnecessary diagnostic and therapeutic procedures such as coronary arteriography and coronary artery bypass grafting (CABG). A number of studies have demonstrated the routine use of ⁸²Rb PET for myocardial perfusion imaging (MPI) at a cost similar to that for ^{99m}Tc SPECT [99]. With the advent of hybrid PET/CT driven by technological advances in medical imaging, there are a growing number of centers using cardiac PET with ⁸²Rb [100] and this has greatly contributed to increasing acceptance of PET for clinical practice. This trend is expected to continue in the foreseeable future.

⁴⁴Ti/⁴⁴Sc generator

The ⁴⁴Ti radionuclide ($t_{_{1/2}} = 60.6$ y) decays by electron capture (EC) to ⁴⁴Sc ($t_{_{1/2}} = 3.97$ h) which subsequently decays to stable ⁴⁴Ca. ⁴⁴Sc is also a positron emitter ($\beta_{branch}^+ = 94.3$ %) and is a valuable alternative to ⁶⁸Ga as a matched pair for radiotherapy. The use of ⁴⁴Sc with a half-life more than 3 times longer than that of ⁶⁸Ga ($t_{_{1/2}} = 68$ min) makes it a valuable alternative for diagnostic and dosimetry purposes. Decay characteristic of ⁴⁴Ti/⁴⁴Sc system is shown in **Figure 8**.

The 3.97-hour half-life of ^{44}Sc is 3 times longer than that of ^{68}Ga and therefore it may be a useful alternative not only for diagnostic purposes but also for dosimetry studies and further therapy planning with the use peptides labeled with the β -emitting radionuclide such as ^{177}Lu or ^{90}Y as radiotherapeutic agents.

The 3.92-hour half-life of 44 Sc provides the ability to elute at 4 h interval with about a 50% elution yield. Hence, a 370 MBq (10 mCi) generator should be able to give 148-185 MBq (4-5 mCi) dose every four hours.

Laboratory	Target	Irradiation conditions	Typical batch yield at EOB (GBq)
Brookhaven National Lab (BNL), NY, USA	RbCl pressed pellet in inconel	2 targets, 93-70 & 64-41 MeV	220-300 (5.9-8.1Ci)
Los Alamos National Lab (LANL), NM, USA	RbCl cast puck in inconel	2 targets, 97-71 & 65-45 MeV	300-450
Institute for Nuclear Research (INR), Troitsk Russia	Rb metal in stainless steel	100-40 MeV	120-220
iThemba Labs, Faure, S. Africa	Rb metal in stainless steel	66-44 MeV	100
Arronax, Nantes, France	RbCl pressed pellet in stainless steel	8 thin targets, 69-44 MeV	80-90
Nordion/TRIUMF, Vancouver Canada	Rb metal in stainless	60-48 MeV	60-100

Table 8. Production details from the current suppliers of ⁸²Sr

	3 1				
Depation		Yield [mCi (µA.h) ⁻¹]			
Reaction	Energy Range (MeV)	⁷² Se	⁷³ Se	⁷⁵ Se	
⁷⁵ As (p, 4n)	45-24	0.21	57	0.03	
⁷⁵ As (d, 5n)	56-26	0.15	32	0.08	
^{nat} Ge (³ He, xn)	36	0.08	NR	NR	
⁷⁰ Ge (³ He, xn)	36	0.22	NR	NR	
Cu ₃ ^{nat} Ge (⁴ He, xn)	28-10	0.0005	0.09	0.0004	
Cu ₃ ⁷⁰ Ge (⁴ He, xn)	28-10	0.002	0.4	0.0003	
K ^{nat} Br (p, x)	100-69	0.10	7.18	0.12	
K ^{nat} Br (p, x)	62-42	0.05	2.36	0.03	
Rb ^{nat} Br (p, x)	800	3.03	30.3	3.03	

NR: Not reported.

The 60.6 years half-life of ⁴⁴Ti offers long-term application without generator replacement.

Table 9. Possible ways for the production of ⁷²Se

Despite its high β^+ -branching (94.3%), ⁴⁴Sc shows an additional 99.9% photon emission of 1157 keV which is expected to be useful for nuclear medicine imaging using β - γ coincidences.

The chemistry of ⁴⁴Sc is similar to that of the lanthanides, and the "lanthanide like" elements. Hence chelators developed for the complexation of ¹⁷⁷Lu and ⁹⁰Y such as DOTA and DTPA analogs can also be used for the complexation of ⁴⁴Sc.

While ⁶⁸Ga labeled compounds have significant clinical potential and present a convenient, low-cost alternative to cyclotron-produced PET radiopharmaceuticals, the 68 min physical half-life of ⁶⁸Ga limits the spectrum of clinical applications of ⁶⁸Ga-labelled radiodiagnostics. In addition, ⁶⁸Ga-labelled analogues of endora-diotherapeuticals with longer biological half-lives, such as ⁹⁰Y- or ¹⁷⁷Lu-labeled peptides and proteins, cannot be used to determine individual radiation dosimetry directly. In this context, the notion of using generator-derived positron

emitters with longer physical half-lives, such as ^{44}Sc (t_{_{4_2}} = 3.97 h), is deemed worthy of consideration.

While the outlook of ⁴⁴Ti/⁴⁴Sc generator is promising, development of a generator amenable for clinical use will not be a trivial process and poses formidable scientific and technical challenges, such as effective separation strategies of availing optimum ⁴⁴Sc elution yields and low ⁴⁴Ti breakthrough, and selection of an appropriate eluent to obtain ⁴⁴Sc in a chemical form suitable for subsequent radiolabeling

(i.e. high radioactive concentration (RAC), low pH, free from metallic impurities etc.). With an aim to tap the potential of ⁴⁴Ti/⁴⁴Sc generator to avail ⁴⁴Sc of requisite purity, several strategies have been explored [101] among which the method reported by Filosofov *et al.* [102] merits attention. This 185 MBq (5 mCi) ⁴⁴Ti/⁴⁴Sc generator system has been studied in detail and has been successfully exploited for preclinical investigations.

The generator consists of a chromatography column containing Bio-Rad AG1 × 8 (200-400 mesh, Br⁻-form), an anion-exchange resin, on to which purified ⁴⁴Ti dissolved in 20 mL of 0.1 M $H_2C_2O_4$ was loaded. ⁴⁴Sc could be eluted with 20 mL of 0.005M H₂C₂O₄/0.07 M HCl solution with > 97% elution yield and low 44Ti breakthrough (5×10^{-5}) . The low RAC and the chemical composition of the generator eluate emerged as the major hindrances that thwarted efforts for fast, reliable and quantitative radiolabeling of nanomolar concentration of precursors. In order to enhance the RAC, fractionation of the elution of ⁴⁴Sc was resorted to, but met with limited success, as the concentration of hydrochloric acid was too high for realizing direct radiolabeling.

With to the goal of enhancing RAC, reducing HCl concentration, and removing the oxalate anions, the prospect of using post elution processing was considered to be reliable and subsequently pursued [103]. In this procedure 0.005 M H_2C_2O_4 / 0.07 M HCl solution was passed through the ⁴⁴Ti/⁴⁴Sc generator to elute ⁴⁴Sc adsorbs, which was again passed online through a small cation exchange column to retain ⁴⁴Sc. The cation-exchange column was washed with 2-4 mL of H₂O to remove the remaining traces of the initial eluate solution. ⁴⁴Sc retained by the cation exchange column was then eluted with 3 mL of 0.25 M ammonium acetate buffer, pH = 4.0. Reconditioning of the cation exchange cartridge was performed by passing 1 mL of 4 M HCl and finally by 1 mL H₂O. A schematic representation of the procedure is depicted in Figure 9. A 185 MBq (5 mCi) generator provides about 170 MBg of ⁴⁴Sc after direct elution and about 150 MBq following online post-processing.

Scandium is chemically similar to Y^{3+} and lanthanides. While the ionic radius of Sc^{3+} (74.5 pm) is smaller than that of lanthanides, it is larger than any of the M^{3+} cations formed by the 3d transition metals. It has been established that Sc^{3+} in solution has a tendency to forms DOTA complexes with coordination number 8 and 9 similar to that Y^{3+} and Lu^{3+} , whereas Ga^{3+} forms 6 coordinated octahedral complexes [104]. $^{44}Sc^{3+}$ as a metallic cation is suitable for complexation with chelators alone or with BFCs (DOTA, DTPA, NOTA, etc) conjugated to peptides or other molecular targeting vectors similar to the currently-used, trivalent radionuclides such as ^{68}Ga , ^{111}In or ^{90}Y and ^{177}Lu [105].

Recent studies on a series of polyamino-polycarboxylate ligands has shown that Sc-8dentate ethylene glycol-bis (2-aminoethylether)-N,N,N',N'-tetraacetic acid (EGTA) is a promising moiety for coupling ⁴⁷Sc to proteins [106]. It has been demonstrated that DOTA is the most preferred chelator for ⁴⁴Sc(III) among a series of macrocyclic ligands [52]. All 4 amine and carboxyl groups of a DOTA chelator contribute to the complexation of ⁴⁴Sc(III), resulting in a stability constant comparable to those of Y(III) and lanthanides [52]. Also, ⁴⁴Sc-DOTATATE has similar lipophilicity to that of ¹⁷⁷Lu- and ⁹⁰Y-DOTATATE [52]. Due to chemical similarity of Sc3+ with Lu3+ and Y³⁺ cations, it is anticipated that ⁴⁴Sc-DOTA bioconjugates will have similar in vivo characteristics (*i.e.* receptor affinity, kidney clearance) to the ¹⁷⁷Lu- and ⁹⁰Y-conjugates currently used in therapy.

The ⁴⁴Sc-DOTA-conjugated tumor targeting vectors such as somatostatin analogs investigated so far are stable *in vitro* and *in vivo* [107, 108], and the data obtained reveal pharmacological parameters which are adequate for performing molecular imaging. Thus, ⁴⁴Sc seems poised to become an alternative to ⁶⁸Ga for imaging and dosimetry before ¹⁷⁷Lu- or ⁹⁰Y-radionuclide tumor therapy [101].

⁶²Zn/⁶²Cu generator

⁶²Cu is a PET radionuclide with a half-life of 9.74 minutes, and decays 97.8% by β⁺ and 2.2% by EC to the ground state of stable ⁶²Ni, with 99.6% of the β⁺ emissions having a 2.9 MeV positron E_{max} [109].

The 9.7 min half-life of ⁶²Cu is long enough to facilitate radiopharmaceutical synthesis procedures and for image acquisition, while at the same time, it is short enough to afford favorable dosimetry.

The 9.7 min half-life of 62 Cu enables serial imaging procedures in the same clinical visit without interference of 62 Cu background activity from previous injections.

The versatile chemistry of copper provides the ability to prepare a wide range of ⁶²Cu-labeled PET-radiopharmaceuticals for a variety of imaging applications.

The short half-life and high positron decay fraction (99.6%) of ⁶²Cu make it suitable for characterizing the faster kinetics of smaller tracer molecules, such as Cu-PTSM or Cu-ATSM.

⁶²Cu could be used as a PET-surrogate to therapeutic isotope ⁶⁷Cu.

While the 9.7 min half-life of ⁶²Cu restricts imaging to less than an hour following injection, this has the advantage of allowing several repeat scans in quick succession, which can be used to evaluate the response to a number of physiological or pharmacological stimuli.

The availability of 62 Zn/ 62 Cu generators is often the ultimate limiting factor in realizing the benefit of 62 Cu-radiopharmaceuticals. Evolution and continued success of ⁶²Cu in PET has been, in large part, due to development of the ⁶²Zn/⁶²Cu generator (**Figure 10**). Advances in the development of ⁶²Zn/⁶²Cu generator have contributed substantially to the deployment of ⁶²Cu-radiopharmaceuticals in clinical PET.

Robinson et al. [110] prepared a ⁶²Zn/⁶²Cu generator system based on "Dowex 1-X10" resin of 200-400 mesh [0.7 cm (Φ) × 8.0 cm (I)] from which ⁶²Cu could be eluted with 3.5 mL of 0.1 N HCl containing 100 mg/mL NaCl and 1 µg/mL CuCl_o, with 85% elution yield and with ⁶²Zn breakthrough less than 0.001%. Fujibayashi et al. [111] has developed a generator based on a cation exchanger (CG-120, Amberlite), from which ⁶²Cu is eluted in a glycine complex. While this system offers 70% elution efficiency, the ⁶²Zn breakthrough was significantly high (2.2%). Zweit et al. [112] have developed a 62Zn/62Cu generator based on anion-exchange resin from which 62Cu could be eluted in a solution of 0.3 M HCI/40% ethanol with greater than 90% elution yield in a 3-mL volume and with low 62 Zn breakthrough (< 3 × 10⁻⁷%). Bormans et al. [113] have developed an improved 62Cu-generator system based on Dowex 1 × 16, 200-400 mesh, from which ⁶²Cu is eluted using a solution containing 1.7 M in NaCl/0.1 M in HCl. Fukumura et al. prepared an improved ⁶²Zn/⁶²Cu using Sep-Pak plus CM cartridge, from which 62Cu was eluted with high elution efficiency (approximately 96%) using a small volume (ca. 3 mL) of a 200-mM glycine solution with a very low breakthrough of ⁶²Zn (< 0.1%) [114]. El-Azony has reported a method based on a De-Acidite FF anion exchanger [115], from which ⁶²Cu is eluted using a solution of 0.2 M HCI-60% acetone with an elution efficiency of 92.5%. As the 0.2 M HCI-60% acetone medium is unsuitable for medical use, it was evaporated and reconstituted in a chemical form suitable for radiolabeling.

The chemistry of copper is restricted to two principle oxidation states (I and II), and the relatively simple coordination and redox chemistry of copper is well documented [116]. Copper salts generally exists as $[Cu(OH)_6]^{2+}$ in aqueous solution. Compounds of the Cu(I) oxidation state are unstable in aqueous solution and readily oxidize to Cu(II) to form 4, 5, or 6 coordination bonds with ligands. The Cu(II) oxidation state has a pronounced tendency to form coor-

dination complexes with ligands that contain electron donor atoms, such as, N and S. Complex formation with chelating agents occurs at pH < 7 since formation of insoluble $Cu(OH)_2$ at higher pH is a major concern [117]. The *in vivo* stability of the radio-copper complex is a critical factor for designing ⁶²Cu-radiopharmaceuticals. Ligands that can form kinetically inert Cu(II) complexes are ideal since this is more significant than thermodynamic stability.

⁶²Cu(II) diacetyl-bis(N4-methylthiosemicarbazone), or 62Cu-ATSM, has emerged as an effective agent to image tissue hypoxia [118-120]. ⁶²Cu-ATSM is a neutral, lipophilic compound that can be readily diffuses into all the cells with perfusion. However, its structure changes due to the reduction of Cu²⁺ cation to Cu⁺ cation in hypoxic conditions resulting in accumulation in the hypoxic cell. The mechanism of ⁶²Cu-ATSM retention in hypoxic tissues is ascribed mainly due to the low oxygen tensions and the subsequent altered redox environment of hypoxic tumors. Results from in vivo studies inferred that tissue Cu-ATSM uptake is dependent on oxygen concentration [121]. ⁶²Cu-ATSM has shown heterogeneous uptake in tumors with homogeneous perfusion images, strongly suggesting uptake reflecting hypoxic heterogeneity. Results from several studies demonstrated the utility of Cu-ATSM PET hypoxic imaging with improved prognosis and effectiveness of radiotherapy [122-128].

⁶²Cu(II) pyruvaldehyde bis(N4-methylthiosemicarbazone), or ⁶²Cu-PTSM, is a promising tracer for myocardial, cerebral, renal, and tumor perfusion agents, and has attracted considerable interest [128-132]. 62Cu-PTSM demonstrated a high, first-pass extraction along with prolonged tissue retention due to intracellular reductive decomposition of the lipophilc Cu-PTSM complex to liberate the ⁶²Cu ion, with negligible washout from the myocardium cell, suggesting the ability of ⁶²Cu-PTSM to quantify blood flow in organs such as the heart, brain, and kidneys [130, 133, 134]. ⁶²Cu-PTSM is reduced readily by the mitochondria, and is retained in most tissues, and thus can be used as a marker of tissue perfusion [135]. While Cu-ATSM and Cu-PTSM share the same molecular structural backbone, they have significantly different properties when utilized in PET imaging due to the presence of an additional methyl group on the ligand backbone of Cu-ATSM [136].

Another agent in the bis (thiosemicarbazone) family which merits attention is the ⁶²Cu ethylglyoxal bis (thiosemicarbazone), or Cu-ETS, which is under investigation in human studies. The ⁶²Cu-ETS radiopharmaceutical exhibits properties similar to ⁶²Cu-PTSM *in vivo* due to its structural similarity with ⁶²Cu-PTSM, while avoiding limitations imposed by human albumin binding. This agent has shown more linear uptake at high blood flow rates and thus may provide a superior PET perfusion tracer for applications such as myocardial perfusion and renal blood flow measurements [137-139].

As blood flow and hypoxia are complex, interrelated factors of physiology, the prospect of using dual-tracer studies with ⁶²Cu-PTSM (blood flow) and ⁶²Cu-ATSM (hypoxia) affords the potential for characterizing both tumor blood flow and hypoxia in a single scan [120, 140, 141]. In this modality, dynamic signals from overlapping tracers with staggered injections are recovered using differences in tracer kinetics and decay. The multi-tracer PET imaging paradigm has great potential to avail complementary information that would help for treatment selection, planning, and early response monitoring.

Proportional Technologies, Inc. USA has developed a micro 62 Zn/ 62 Cu generator to avail 62 Cu tracer at very high isotonic concentration amenable for clinical use. Proportional Technologies, Inc. has also developed include H₂PTSM, H₂ETS and H₂TASM kits available for both non-human studies and clinical investigation. Components in each injectable solution are described in **Table 6.**

Despite the many advantages, the utility of ⁶²Zn/⁶²Cu generators comes with a few limitations. Currently, ⁶²Zn/⁶²Cu generators are procured on an as-needed basis as they are relatively expensive and have a shelf life of 1-2 days, and patients must be scheduled several days in advance of imaging. The cost of ⁶²Zn/⁶²Cu generator can be significantly off set if demand for the ⁶²Cu compounds increases to the point at which generators can be produced for routine clinical use and the scheduling of several patients for ⁶²Cu PET studies on the day of generator delivery. One generator will support 20 or more dose preparations during 1 day of use, and these doses can be produced at frequent intervals of 30-45 minutes between elution. In this context use, of a lyophilized kit formulation seems poised to provide a practical means of producing ⁶²Cu-PTSM, ⁶²Cu-ATSM and 62Cu-ETS on-site and bring substantial benefits at lower cost, as is well known in radiopharmaceutical practice. Additionally, one of two distribution models for the ⁶²Zn/⁶²Cu generator can be considered. Production of ⁶²Zn can be carried out either in or near the 18F radiopharmacies using a 19 MeV cyclotron and ⁶²Zn/⁶²Cu generator can then be delivered using the same local established delivery network already in place for ¹⁸F. Alternatively, the ⁶²Zn parent can be produced using > 25 MeV cyclotrons and loaded into generators which can be shipped to the local radiopharmacies or hospital radiopharmacies.

⁷²Se/⁷²As generator

⁷²As is a positron-emitting arsenic isotope with a half life of 1.08 d, with branching ratio of 88% and $E_{\beta+,max} = 2.49$ MeV. The chemical properties of ⁷²As are amenable for the preparation of wide range of ⁷²As-labelled PET radiopharmaceuticals. Interest in the use of a ⁷²Se/⁷²As generator is primarily attributed to the following:

Its relatively high β^+ abundance and hours-long half-life allows for imaging of slower biological processes.

The relatively long half-life of the parent 72 Se (t_{1/2} = 8.4 d) allows usage of the 72 Se/ 72 As generator for 2-3 weeks.

The long physical half-life of 1.08 d may render ⁷²As a PET radionuclide of choice for the quantitative imaging of biochemical and physiological processes with longer biological half-lives, e.g. immunoimaging and receptor mapping.

The longer half-life allows for more elaborate chemical modification and labeling methodologies.

The versatile chemistry of arsenic would permit the radiolabelling of a broad spectrum of potentially valuable pharmaceuticals.

⁷²Se/⁷²As generators offer the prospect of availing ⁷²As at the hospital radiopharmacies without dependence on an accelerator facility in close proximity.

Recognizing the potentially important role of the ⁷²Se/⁷²As generator in NM, a wide range of adaptive separation strategies with an aim to obtain ⁷²As in a suitable chemical form and of requisite purity have been reported. An ionexchange chromatography ⁷²Se/⁷²As generator based on Dowex 50 was developed by Al-Kouraishi and Boswell [142]. In this process, cold selenous acid was added to ⁷²Se solution containing sulfur to reduce Se which was obtained as a precipitate. This was then centrifuged, washed with deionized water, loaded onto the chromatography column containing cation exchanger column and held for 90 hours for the in-growth of the ⁷²As daughter. It is possible to elute ⁷²As from the column using deionized water with a 70% elution yield. Another generator system developed at Los Alamos National Laboratory was based on the addition of selenium carrier in the form of selenic acid and hydrazine for cyclic reduction of selenium to Se(0) followed by separation of ⁷²As by filtration with subsequent oxidative dissolution of Se(0) using H_0O_0 prior to each separation cycle [143]. A ⁷²Se/⁷²As generator based on an electrochemical process with selective deposition of ⁷²Se on Pt electrodes as Cu⁷²Se were also reported [144]. Jennewein et al. reported a ⁷²Se/⁷²As generator based on distillation technique [145]. In this work, ⁷²As produced by radioactive decay of ⁷²Se was distilled off as the volatile AsCl, at 105°C for 10 minutes in a gaseous HCl stream, and collected on a charcoal filter with a radiochemical yield > 99%[145]. The same group of authors have reported a system based on a polystyrene solid support [Varian ENV solid phase extraction cartridge] to adsorb elemental ⁷²Se (0) [146]. The in-grown ⁷²As was then eluted using 2 mL of concentrated HF with a 50% elution yield [146]. Chajduk et al. have proposed a method based on extraction chromatography for isolation of arsenic from selenium [147]. In this method aromatic o-diamine extractant was impregnated into a polystyrene column matrix. The extractant selectively retains ⁷²Se from which ⁷²As could be selective eluted using 0.9% NaCl solution. The proposed separation procedure assures > 95 % elution yield with low (< 0.01%) ⁷²Se breakthrough [147]. A ⁷²Se/⁷²As RNG system utilizing chelation and liquid-liquid extraction has been reported [148]. Despite the impressive progress, the promise to develop a commercial, clinical scale ⁷²Se/⁷²As generator amenable for use in a hospital radiopharmacy has not been fulfilled.

Arsenic forms stable covalent bonds with carbon and sulphur and can substitute phosphorus in certain compounds with minimal alteration of the biologically activities of the parent molecule. Jennewein et al. has reported a method to radiolabel 74As with N-succimidyl Sacetylthiacetate (SATA) derivitized bavituximab [149]. Arsenic has a high affinity to sulfur and As is able to bind covalently to sulfhydryl groups. In antibodies, the sulfur moieties are mainly associated with dithiol bridges. To increase the number of free thiols, conscious modification of antibodies with SATA (N-succinimidyl S-acetylthioacetate) seemed sagacious and was pursued [149]. It has been demonstrated that ⁷⁴As labeled bavituximab-SATAconjugate retained its immunoreactivity and exhibits favorable in vitro stability in fetal bovine serum. Biodistribution studies carried out with Dunning prostate R3327-AT1 tumor bearing rats reveal that tumors were clearly visible after 48 hours, exhibiting 8-fold higher uptake at 72 hours as compared to the control antibody [149, 150].

It is clear that ⁷²As based radiopharmaceuticals for clinical PET are still in their infancy and their utility is limited by the commercial unavailability of ⁷²Se/⁷²As generators as well as end-user radiolabeling techniques. Additional preclinical and clinical studies are warranted to exploit the full potential of ⁷²As. Having established their importance in NM, the exploration of ⁷²Aslabeled PET tracers will continue to rise.

¹¹⁰Sn/^{110m}In generator

^{110m}In is a positron-emitting isotope of In, with a half-life of 69.1 min and branching ratio of 62% with $E_{\beta+,max} = 2.26$ *MeV*. This radioisotope is of interest as it would provide the scope of preparation of radiolabeled agents (analogous to SPECT tracers) to be quantified with PET and allow for sequential studies to be performed within a relatively short time. Potential use includes preparation of ^{110m}In labeled leuko-cytes for infection/inflammation imaging [151] and ^{110m}In-octreotide in clinical imaging of neuroendocrine tumors (NET) [152]. The prospect of using ^{110m}In-labeled octreotide seemed to be an interesting proposition as it can better detect small tumors and might be able to more

accurately quantify tumor uptake with PET than can ¹¹¹In-labelled octreotide and SPECT. Due to the short half-life of the 110mIn, use of the Octreoscan kit has been embraced by investigators [152]. Since octreotide is commercially available in the Octreoscan kit (Mallinckrodt Medical, St. Louis), an optimized procedure for labeling with ^{110m}In could be routinely used, solving the problems associated with GMP production of the peptide-chelator conjugate. The 69-min half-life of ^{110m}In matches the rapid kinetics of the octreotide well. Apart from enabling detection of smaller tumors, ^{110m}Inoctreotide-PET can also provide quantitative information about receptor kinetics and concentrations with better temporal and spatial resolution than ¹¹¹In-octreotide-SPECT.

This generator was prepared using Kieselgel 40 (silica gel) conditioned with 0.02 M HCl [153]. ¹¹⁰Sn was loaded in the column at 0.02 M HCl and elution of ^{110m}In was performed using 0.02 M HCl [153]. Owing to the relatively short half-life of the parent, ¹¹⁰Sn ($t_{_{1/2}}$ = 4.9 h), this generator has a limited shelf-life of 10-12 hours. The generator was eluted at 2 h interval with > 90% elution yield of ^{110m}In and the level of ¹¹⁰Sn breakthrough was < 0.003%.

⁵²Fe/^{52m}Mn generator

^{52m}Mn is a positron-emitting isotope which decays with a branching ratio of 98.3% with a 21.1 min half-life and with $E_{\beta+, max} = 2.631 MeV$. In addition to the annihilation radiation, ^{52m}Mn also emits a I434-keV gamma ray (98.3%). The remainder of the decay is by isomeric transition to ⁵²Mn, which has a 5.59-day half-life. This PET isotope has been considered as a as a potential agent for myocardial imaging studies owing to the possibility to perform a number of studies in the course of a day [154-156]. The tremendous prospect associated with the use of ⁵²Fe/^{52m}Mn generators in NM has led to the development of a number of strategies [155-159]. Due to the short half life of ⁵²Fe ($t_{1/2}$ = 8.27 h) breakthrough up to 1% would not significantly alter the efficacy of the ^{52m}Mn eluted. This generator has a very limited shelf life due to the short half life of ⁵²Fe, the parent radioisotope.

Despite the favorable physical properties of ^{52m}Mn²⁺ as an agent for myocardial imaging studies, results from pig model studies con-

cluded that ^{52m}Mn allows the qualitative assessment of myocardial perfusion but does not meet the requirements of a quantitative myocardial perfusion agent [160]. The ⁵²Fe/^{52m}Mn generator presents no chemical or practical advantages over the ⁶²Zn/⁶²Cu generator, which has similar parent and daughter half-lives.

¹²²Xe/¹²²I generator

The use of $^{122}\text{Xe}~(t_{_{1\!/_2}}\text{=}20.1~\text{h})/^{122}\text{I}~(t_{_{1\!/_2}}\text{=}3.6~\text{min};$ β^+ = 77%, EC = 23%, E_{$\beta^+, max}$ = 3.1 MeV) genera-</sub> tor system as a means of availing ¹²²I for PET studies has caught the attention of NM clinicians and evoked excitement among radiopharmaceutical scientists. ¹²²I is unique among all generator produced PET radionuclides in that it is not a metallic element and it is availed from a parent nuclide that is an inert gas. Due to the gaseous nature of the parent radionuclide. the customary use of column chromatography technique for the separation of daughter radionuclide will not be appropriate for this generator. In view of this, suitable alternative means for this separation have been described [161, 162].

While the 3.6-min half-life of ¹²²I, as well as the versatile chemistry of iodine, are appealing for many PET applications, the chemistry of iodine is not particularly amenable to the rapid synthetic chemistry which is dictated by the physical half-life of this label. Nevertheless, a few attempts have already been made. Potential utility of ¹²²I labeled lodoperidol (IP), an iodinated analogue of the antipsychotic drug haloperiodol, as a cerebral blood flow radiopharmaceutical for PET has been evaluated and showed promising results [163]. This study indicates that this class of compounds holds promise for development as perfusion radiopharmaceuticals. 122 has been successfully incorporated into an amphetamine analog, 2,4-dimethoxy-N,N-dimethyl-5-[1221]iodophenylisopropylamine (5-[122I]-2,4-DNNA), using a remote synthesis protocol and studied in dog models as a quantitative cerebral blood flow agent for PET [164].

In vivo generators

This concept essentially consists of labeling of molecular carriers (complexes, peptides, monoclonal antibodies and their fragments, etc.)

with intermediate half-life generator parents, which continuously decay and generate shorter half-life daughter radionuclides much more than the parent [165-167]. It is pertinent to note that the chemical binding of the daughter nuclide must be analogous to the parent one to preclude the release of daughter radionuclide from the original position. Owing to the decay kinetics, the daughter nucleus experiences some recoil. However, this is assumed to be negligible. This is the reverse of the usual use of a generator, where the daughter is initially separated from the parent prior to use. By contrast, with this strategy the daughter is removed from the parent using a chemical separation technique and the parent is then attached to tissue-specific therapeutic agents (complexes, chelate, peptides, monoclonal antibodies and their fragments, etc.) to be administered. While this innovative paradigm has been practiced in radionuclide therapy to minimize the radiation exposure to non-target tissues, this could be extended to imaging agents [168]. Such a strategy seemed appealing, as it has the potential for quantitative PET to inform on a more personalized treatment strategy, since the same carrier moiety can be radiolabeled with a suitable parent/daughter pair to deliver therapeutic or diagnostic imaging radionuclides. In this context two generators that merit attention for the "Theranostics" concept are: ¹⁴⁰Nd/¹⁴⁰Pr and ¹³⁴Ce/¹³⁴La generators [168].

¹⁴⁰Nd/¹⁴⁰Pr generator

¹⁴⁰Nd decays 100% by EC with a half-life of 3.37 days to produce a short-lived positron emitter ¹⁴⁰Pr which decays to stable ¹⁴⁰Ce by positron emission with a branching ratio of 49 %, with $E_{\beta+, max} = 2.4$ *MeV* and with a 3.4 min half-life. This system shows potential as an RNG or as an *in vivo* generator system for PET [168].

A ¹⁴⁰Nd/¹⁴⁰Pr RNG system based on physicochemical transitions (hot-atom effects) of the daughter ¹⁴⁰Pr following the electron capture process of ¹⁴⁰Nd has been developed [169]. In this process ¹⁴⁰Nd(III), in the form of ¹⁴⁰Nd-DOTAconjugated complexes, was quantitatively retained on a cation-exchange resin (Bio-Rad AG 50W-X8, 200-400 mesh, in hydrogen form). ¹⁴⁰Pr was eluted using 10⁻³ M DTPA in > 93% yield with negligible levels of ¹⁴⁰Nd breakthrough.

¹³⁴Ce/¹³⁴La generator

¹³⁴Ce is an Auger electron-emitting radionuclide which decays to ¹³⁴La with a half-life of 3.16 d and the daughter nuclide ¹³⁴La (t_{14} = 6.45 min) is a positron and Auger-electron emitter. While ¹³⁴La was proposed as PET perfusion imaging agent [170], real practical applications have not vet been described. Both the low-energy Auger electrons, emitted in the ¹³⁴Ce and ¹³⁴La decays, as well as the high energy positrons emitted in the ¹³⁴La decay, can be conscientiously exploited for radionuclide therapy. The kinetics of the positron-emitting daughter nuclide can be measured with PET and used for dosimetry. ¹³⁴Ce/¹³⁴La absorbed doses to single cells were reported to be higher than absorbed doses from ⁹⁰Y and ¹¹¹In [170]. Further investigations are warranted to realize the full potential of this system.

Production of parent radionuclide

Any advancement in PET RNGs will be largely dependent upon the availability of required quantity and quality of parent radionuclides. The major challenge for the production of parent radionuclide requires selection of the most economical processes from the pool of available options. Most of the parent radionuclides used in RNGs are produced using solid targets of metal-based materials (e.g. elemental metal or metal oxide) with the target either mounted to the cyclotron (under vacuum) or external to the cyclotron/accelerator. In each case, the target system needs to tolerate heat generated from the particle bombardment. Depending on the energy and current of the beam, the cooling systems can be quite elaborate. As the number of nuclear reactions in a given target material can be potentially large, one needs to design the target so that the preferred nuclear reaction dominates. This may be achieved in a number of ways:

Choice of beam incident particle [i.e. p, d, or α] energy of the beam, target material (i.e. natural or enriched isotope of target element).

Routine production of parent radionuclides for RNGs should meet the following requirements:

The production method, involving the bombardment as well as the radiochemical separation procedure for the isolation radionuclide of interest, must be economically viable.

Ability to isolate the radionuclide of interest in high specific activity, with acceptable radionuclidic, radiochemical and chemical impurities.

The radiochemical separation method selected must be as simple as possible to facilitate remote operation/automation within a hot cell to minimize radiation exposure to the operator.

The following section provides an overview of the issues associated with production of parent radionuclides which should be considered to identify recent advances in this field.

⁶⁸Ge

Two nuclear reactions have been utilized for the routine production of ⁶⁸Ge, namely, by ⁶⁶Zn(α,2n)⁶⁸Ge (⁶⁶Zn natural abundance being 27.8%) giving a yield of up to $2 \mu \text{Ci.}(\mu \text{A.h})^{-1}$ (yield per µA beam current calculated for 1 h irradiation) at 35 MeV beam energy, or by ⁶⁹Ga(p,2n)⁶⁸Ge (⁶⁹Ga natural abundance being 60%) giving a yield of up to 20 μ Ci.(μ A.h)⁻¹ at 23 MeV beam energy [171]. Of the two reactions, ⁶⁹Ga(p,2n)⁶⁸Ge has been regarded as the reaction of choice due to the higher yields obtained and ease of chemical separation. In this method, only two elements need to be separated from each other, where with the other reaction, a third element (Zn) needs to be taken into consideration [172].

When selecting the target, chemical, mechanical and thermal properties, and corrosion and radiation resistance need to be considered since the target will be exposed to high current irradiations and the power dissipated in the targets reaches values of about 300-1000 W or more. Target materials used for the production of 68Ge include Ga2O2, Ga4Ni, GaAg and Ga metal. The use of encapsulated Ga203 with high-current proton beams is precluded as the oxide changes from a hexagonal α -form to a monoclinic β-form at about 600°C accompanied by a volume increase, which leads to capsule rupture. While the use of alloys as target material (Ga₄Ni and GaAg) has the advantage of attaining good thermal conductivity, the requirement of an elaborated chemical separation procedure to make ⁶⁸Ge free from coproduced impurities discourage their use. The

prospect of using Ga_2O_3 is prohibited by the difficulties encountered during the dissolution of irradiated target. In this regard, the idea of using Ga metal seemed attractive. As Ga metal is corrosive to Al, it is essential that it be encapsulated, using Nb for example, as Ga would not react with this encapsulation.

In order to avail satisfactory batch yields, proton energy > 20 MeV, high-current accelerators on the order of the mA's and long irradiation periods of several days duration are required. Hence there are very few suppliers of ⁶⁸Ge, despite huge demands. Four major centers which produce 68Ge currently are: iThemba laboratories (South Africa), Brookhaven National Laboratory (USA), Los Alamos National Laboratory (USA) and Cyclotron Co Ltd (Obninsk, Russia) [171]. These centers report on production capacities of about 18.5 to 74 GBg (0.5 to 2 Ci) of ⁶⁸Ge per batch. Separation techniques used to isolate micro quantities of no-carriedadded (NCA) 68Ge from macro amount of irradiated target range from solvent extraction [173-176] and ion exchange chromatography, to using organic [177] and inorganic [172, 178, 179] materials, have been employed.

At BNL, production is carried out using ^{nat}Ga targets with ~45 MeV protons. For a typical batch production, 81 g of natGa metal encapsulated in a Nb container is used and irradiation was carried for period of 4 weeks (0.52 MBq. (µAh)⁻¹. ⁶⁸Ge is recovered from the target by extraction into 4 N HCl and 30% H₂O₂ after two weeks cooling. Further purification is achieved by solvent extraction using carbon tetrachloride and back-extraction of 68Ge into H₂O. An overall recovery yields > 85% having a batch yields of 33.3-51.8 GBq (900-1400 mCi) with radionuclidic purity > 99.9%, and activity concentrations > 3.15 GBq/mL (85 mCi/mL) have been reported [180]. At the Los Alamos National Laboratory (LANL), 100 MeV protons are used [181]. For a typical batch production, 4 g of ^{nat}Ga metal encapsulated in a Nb container is used and irradiation was carried for period 16-20 days (1.18 MBq /µAh). The batch yield at end of bombardment (EOB) is about 55-70 GBq. Chemical processing is carried out after two weeks of cooling using solvent extraction with CCI, and re-extraction of 68Ge into water followed by an ion exchange purification step using alumina [182]. At the iThemba Laboratories, a 66 MeV proton beam is used. For a typical batch production, 5 g of target with a production rate 1.18 MBq (0.032 mCi). (μ Ah)⁻¹ is used. The radionuclidic purity of the processed ⁶⁸Ge is > 99.9% and the final product contains < 1 μ g of Ga per 37 MBq (1 mCi) of ⁶⁸Ge [183]. At the Cyclotron Co. Ltd. in Obninsk, Russian Federation, Ga-Ni alloy prepared on Cu backings is used as the target material. Irradiations are performed at a high proton beam intensity of several hundred microamperes of 23 MeV protons. ⁶⁸Ge of high specific activity [> 74 GBq (> 2 Ci)/mg] and 99.8% radionuclidic purity is obtained [184].

⁸²Sr

⁸²Sr has been produced via several different reaction routes, as shown in Table 7. Each method presents different challenges for targetry, process chemistry, practical yield, radionuclidic purity, and availability. While the use of high energy spallation reactions with Mo or Y metal as the target constitutes a method for the production of ⁸²Sr [185-187], low reaction cross-sections and the requirement of an elaborate radiochemical procedure to isolate ⁸²Sr from other spallation products, as well as the bulk target material, emerged as the major impediments which continue to discourage this method's wide scale adaptability. In an attempt to produce ⁸²Sr, high energy ³He and alpha particle irradiations of natural Kr have also been tried [188-190], but met with limited success due to low yields and the scarcity of operating accelerators capable of accelerating ³He ions to the energies required. In this light, the ^{nat}Rb(p,xn)⁸²Sr reaction seemed most attractive and thus stands as the technique of choice for large scale production [191-193].

The requirement to use high energy proton reactions leads to relatively large government owned accelerator facilities around the world, and poses challenges associated with irradiation scheduling (as these facilities have other missions and do not operate continuously year round) and timely transport of large quantities of radioactivity to the generator production site; these are some of the major road blocks that need to be surmounted for widespread use of ⁸²Sr for medical applications. As the potential for ⁸²Sr/⁸²Rb generators is evident, several commercial companies have started producing

⁸²Sr to meet the demand on a routine basis. Production details adopted by the current suppliers of ⁸²Sr are depicted in **Table 8**.

With the goal of undertaking a large scale production process using the ^{nat}Rb(p,xn)⁸²Sr reaction, both RbCl and Rb metals have been used. The advantages of using RbCl are its commercial availability in high chemical and isotopic purity, resistance to corrosion, ability to be pressed into a pellet at near theoretical density, or melted and cast into a capsule, and easy dissolution of irradiated targets in water as well as common mineral acid. Limitations on the use of RbCl include hygroscopic behavior of the material and low thermal conductivity. The poor thermal conductivity requires careful target design and limits target thickness in order to withstand high beam current.

The use of Rb metal permit permits the use of very thick targets in order to increase batch quantities because of high thermal conductivity. Disadvantages of using Rb metal include commercial non-availability of very high purity material and the serious safety issues in handling metallic Rb, due to its susceptibility to ignite or explode upon exposure to air or moisture, and the subsequent requirement to perform initial irradiated target dissolution within an inert atmosphere (e.g. dry argon) using an anhydrous higher alcohol, typically propan-2-ol.

At BNL and LANL, processing of the irradiated RbCl target was carried out by removing the exterior of the target capsules with acid, followed by dissolution of irradiated RbCl pellet in water. 82Sr2+ was isolated using an ion exchange chromatography separation process using Chelex 100, wherein ⁸²Sr²⁺ is chelated and retained on the resin while Rb⁺ gets eluted. ⁸²Sr²⁺ was eluted from the column using 6 M HCl and further purification of ⁸²Sr²⁺ was carried out using ion exchange chromatography with AG 50W-X8 and Chelex 100 [171, 194]. At iThemba Labs, the irradiated target is dissolved in dilute ammonium chloride solution and 82Sr is separated from the target material using column chromatographic methods with Purolite S950, a macroporous aminophosphonic acid chelating resin. Further purification of ⁸²Sr is carried out following column chromatography technique using AG MP-50 macroporous cation exchange resin. The final ⁸²Sr possesses high

radionuclidic purity with negligible Rb and Fe chemical impurities [194].

Proton irradiated metallic Rb targets from Institute for Nuclear Research (INR) of the Russian Academy of Sciences, Russian Federation, are also processed at Los Alamos National Laboratory (USA) [195], as well as in Institute for Physics and Power Engineering (IPPE), Obninsk, Russia, using a patented method [196]. Pure ⁸²Sr obtained in IPPE is then transported to the Russian Research Center of Radiology and Surgical Technologies (RRCRST), St. Petersburg, for loading the ⁸²Rb-generator. Positron Corporation, USA has entered into a license agreement with INR to produce ⁸²Sr in its proposed 70 MeV cyclotron and using the patented process technology of INR [197]. The rights to this patented technology have been awarded by INR to Positron Corporation.

Over the past decade, the growth of cardiac PET imaging has driven a significant increase in the use of ⁸²Sr/⁸²Rb generators. Currently, the US Department of Energy (DOE) is the only supplier of ⁸²Sr in the US. Positron Corp, U.S., is working with the US FDA for certification so it can begin its own production of ⁸²Sr, through its subsidiary, Manhattan Isotope Technology (MIT), to provide a second source for ⁸²Sr in the USA. In August 2012, MIT submitted its drug master file (DMF) with the US FDA and has begun production of active pharmaceutical ingredient (API) grade ⁸²Sr at its Lubbock, Texas, facility with strontium received from iThemba South Africa.

⁴⁴Ti

⁴⁴Ti/⁴⁴Sc RNGs described in the literature thus far are all based on the ⁴⁴Ti produced by proton irradiation of scandium, by the ⁴⁵Sc(p,2n)⁴⁴Ti reaction. The long half-life of ⁴⁴Ti (60 years) and a low cross-section necessitates the use of high proton flux and long-duration irradiations. The targets are prepared by melting Sc in 150-220 µm thick layers onto copper backings, which themselves are covered by intermediate layers of Ag in order to reduce ⁶⁵Zn coproduction. About 1.5 g Sc targets are subjected to proton currents of ~200 µA for ~10 h at ~30 MeV proton energy. Irradiated Sc target processing was carried out by dissolving the target in 18 mL of 2 M HCl. Separation of ⁴⁴Ti from silver as well as co-produced ⁶⁵Zn and ¹⁰⁹Cd

have been achieved following a multi-step ionexchange chromatography separation, and purification procedures using AG 50W-X8, 200-400 mesh in H⁺-form, as reported [101, 102]. Using the reported procedure, about 185 MBq (5 mCi) of ⁴⁴Ti was successfully recovered.

⁶²Zn

The common method for the production of the 62 Zn parent is nat Cu(p,xn) 62 Zn using intermediate energy (30-60 MeV) proton accelerators [110, 112]. Using low-energy protons (5-14 MeV), 62 Cu can be produced following 62 Ni (p,n) 62 Cu reaction [198, 199]. As reported, the nat Zn (p,x) 62 Zn process can be adopted as a production route if protons of energies higher than 50 MeV are available [200]. The radiochemical processing step involves dissolution of the irradiated target in concentrated HNO₃, reconstitution in 3 M HCl and separation from Cu using ion exchange chromatography (AG1-X8, Clform) [114].

⁷²Se

Spurred by the perceived need to avail ⁷²Se of requisite purity for the development of ⁷²Se/⁷²As generators, a number of production strategies have been described in the literature [142, 146, 201-203] and are elaborated in Table 9. When preparing for the production of ⁷²Se, following one of the different production pathways, a thorough assessment of the production yield, the ease of the required separation chemistry, and the radiochemical purity of the ⁷²Se product needs to be considered. It is pertinent to point out that highest production yield for ⁷²Se is achieved following the proton induced reaction paths using RbBr salt targets, although this production approach is associated with challenges to isolate 72Se owing to the complexity of the product mixture in dissolved solution. While the irradiation of As targets with intermediate energy protons offers the next best yields for ⁷²Se, chemical processing of multi-gram production targets is not only challenging but also associated with radioactive waste disposal issues because of co-production of long lived ⁷⁵Se ($t_{1/2}$ = 120 d).

Among the various possible routes to be considered for the production of ⁷²Se, the use of alkaline bromide targets seems attractive, as they balance a relatively inexpensive and biologically benign target material with sufficiently high yields for the production of ⁷²Se [148]. Other advantages include ease of dissolution of the irradiated target due to high solubility in water, and a reasonably high thermal conductivity. The target processing step involves dissolution of the irradiated target in dilute HCI. Quantitative removal of bromide is achieved by oxidation to elemental bromine, followed by distillation. A liquid-liquid extraction system using dithiocarbamate/ethyl acetate is used to purify Se [148].

⁵²Fe

A number of nuclear reactions using proton, ³He and α particles have been studied for the ⁵²Fe production: ⁵²Cr(³He,3n)⁵²Fe [2O4], ⁵⁰Cr(⁴He,2n)⁵²Fe [2O5], ⁵⁵Mn(p,4n)⁵²Fe [157, 206, 207], ^{nat}Ni(p,x)⁵²Fe [2O7]. While the ⁵⁵Mn (p,4n)⁵²Fe [157, 206-208] route of production using a 65-70 MeV proton beam has been appealing, the beams available from most medical accelerators are below this energy. Therefore, attempts to prepare sufficient quantities of ⁵²Fe following this path have met with limited success. The production of ⁵²Fe by spallation suffers from the requirement of a high energy cyclotron and the coproduction of ⁵⁹Fe, a troublesome contaminant with a 45 day half-life.

The irradiation of Mn with 65-70 MeV protons is expected to produce isotopes of Fe and Mn. and possibly of Cr with reasonable yield, while V and Ti radionuclides will be formed in very small quantities [206]. The irradiated target is dissolved in 6 N HCl containing H₂O₂. In order to avail ⁵²Fe of required quality, these radionuclides were separated using an ion-exchange chromatography method with Dowex-1 × 8, wherein the Fe gets trapped and the Mn, Na, Cr, and AI pass through the resin column with the 6 N HCl loading solution. Fe retained on the column is eluted with 0.5 N HCl. This step is repeated 2-3 times to obtain Fe of required purity [206]. One of the major limitations of this production route is that ⁵²Fe produced is contaminated with ⁵⁵Fe impurity.

The use of the 50 Cr(4 He,2n) 52 Fe nuclear reaction with an enriched 50 Cr target and an irradiation with an ~40 MeV 4 He beam seemed to be appealing option [209] as it precluded the production of the long-lived 59 Fe isotope. However,

it requires the recycling of precious enriched ⁵⁰Cr target in order to make the production cost effective. The irradiated target contains ⁵²Mn, ⁵⁴Mn, ⁵⁶Mn, ⁴⁹Cr, ⁵¹Cr, ⁴⁸V and ⁵⁵Fe radionuclides in addition to the ⁵²Fe. Separation of ⁵²Fe from radioactive contaminants, as well as ⁵⁰Cr, is achieved by solvent extraction of Fe(III) from 8 M HCl using di-isopropyl ether (DIPE) [209].

¹²²Xe

The nuclear reactions used for the production of ¹²²Xe are: ¹²⁰Te(α , 2n)¹²²Xe, ¹²²Te(α , 4n)¹²²Xe, 124 Xe(p, 3n) 122 Cs which decays to 122 Xe, 124 Xe(p, p2n)¹²²Xe and ¹²⁷I(p,6n)¹²²Xe [210]. The proton induced reaction using ¹²⁷I is most abundantly used because the product and reactants are chemically different. The maximum cross-section for the ¹²⁷I(p,6n)¹²²Xe reaction is expected to be in the 72-74 MeV energy range [162, 211]. While the use of a higher energy proton beam results in significantly higher ¹²²Xe yields, it also leads to the production of other radioisotopes of iodine such as ¹²¹I, ¹²³I, ¹²⁵I due to the decay of ¹²¹Xe, ¹²³Xe, ¹²⁵Xe coproduced in the same energy region. With the aim to reduce the radiation dose hazard and to minimize the level of radioiodine isotopic impurities, proton energy in the range of 72-74 MeV energy is desirable. However, ¹²³Xe is also obtained as a byproduct of irradiation of Nal targets due to ¹²⁷I(p, 5n)¹²³Xe reaction with 67.5 MeV proton beams and this is undesirable [211]. ¹²⁴Xe(p, 3n)¹²²Cs which decays to ¹²²Xe, and ¹²⁴Xe(p, p2n)¹²²Xe are not used for production for economic reasons due to the requirement of enriched and expensive target material which cannot be recovered for recycling as the separation of ¹²⁴Xe from the ¹²²Xe is not feasible.

While the use of ¹²⁰Te(α , 2n)¹²²Xe and ¹²²Te(α , 4n)¹²²Xe reactions offer the potential of availing ¹²²Xe, the requirement of 30-40-MeV α beams, expeditious as well as continuous removal of the radioxenons formed using solid/gas separation technique, and low natural abundances of ¹²⁰Te (0.096%) and ¹²²Te (2.60%) have emerged as the major impediments that discourage their large scale adaptability. Use of expensive enriched target material, the need to recover the target, and restricted availability of an accelerator to produce intense 30-40 MeV α beams are other major hurdles that must be tackled while following this route [211].

The production of ¹²²Xe by spallation using Csl, La₂O₃ or BaCO₃ targets and protons of energies from 320 to 590 MeV has also been reported by Peek and Hegedus [212]. Production of many xenon isotopes, such as ¹²⁷Xe, ¹²⁵Xe, ¹²³Xe, ¹²¹Xe, during spallation and the requirement of > 200-MeV proton beams have emerged as the major deterrents that limit the utility of this production pathway.

Radiation safety

Radiation safety is another important part of the RNG development process. Because of the recent advancements in cyclotron targetry, radiochemistry, and separation techniques, new RNGs capable of providing PET radionuclides with different decay characteristics are entering biomedical research and seem poised to enter into clinical practices. Therefore, there is a clear need to reconsider the radiation safety aspects in the development of PET RNGs. Most of the RNGs are appropriately placed in a shielded container, designed to enable elution of the daughter radionuclide. Meeting the regulatory requirements for safe operation is an expensive proposition, but constitutes a necessity. Careful planning with the design engineer, generator architect, and a qualified medical physicist is necessary to produce a cost-effective design while maintaining the radiation safety standard.

With the aim to minimize both manufacturing and transportation expenses, and to make the system portable, it is essential to design an RNG such that the amount of shielding can be kept as small as possible. Adequate shielding is essential to reduce the risk of radiation exposure to workers. The purpose of the RNG shielding is to attenuate radiation by absorption (and to some extent scattering). This protects radiation workers by reducing their exposure. The design of proper shielding requires the use of accurate y-ray dose (C) constants and tenth value layers (TVLs). In order to achieve the greatest possible radiation safety, the amount of radiation exposure from RNGs should be kept below a safe limit following the "ALARA" (As Low As Reasonably Achievable) principle. Based on the type of PET radionuclide, the design and radiation safety features of RNGs need to be adjusted appropriately.

Regulatory need

While the use of PET RNGs is a successful paradigm for ensuring sustained growth and future expansion of PET, their development and the possibility of their clinical translation have been stymied by a number of factors, such as: the complexity of pharmaceutical legislation and regulations, the lengthy process of obtaining a marketing authorization, and, in some cases, a limited return on investment, as well as lack of consensus. Despite the encouraging prospects and the favorable results of many PET RNGs, there is still quite a long way to go before any one of them become standard for widespread use in daily NM routine.

Among many other requirements, such as those relating to establishment of chemical and radionuclidic purity of the PET radionuclide obtained from an RNG, clinical use of RNGs is dependent on the condition that they are manufactured under conditions of good manufacturing practice (GMP). Radionuclides obtained from RNGs are considered as approved pharmaceutical ingredients (API) as they are used as a starting material for the preparation of radiopharmaceuticals for human use [213], and therefore subjected to regulatory approval to ensure both quality and safety. The emphasis on quality is most prominently manifested by the fact that not only do radionuclides obtained from the generators have to meet strict specifications but also the separation processes and associated accessories must fulfill the preset criteria.

Directive 2001/83 of European Union [214] states that all medicinal products must comply with the current standards of GMP. In Europe, these GMP rules have legal character as they are issued by the European Medicines Agency (EMEA) and the DG Enterprise and Industry of the European Commission. They are published online in the Eudralex, a web-based compendium of European pharmaceutical legislation [215]. However, Directive 2001/83, the basis for European GMP as outlined in Eudralex, only has a mandate for the industrial manufacture of medicinal products and does not legally cover the small-scale, patient-specific preparation, e.g., in pharmacies. Similarly, the US FDA approved a set of regulations describing production of PET radiopharmaceuticals according to current good manufacturing practice (cGMP), outlined in the Code of Federal Regulations (21CFR212), [216]. Because of the differences and complexities of the regulatory machinery in different countries, it would be expected that each country is different, and the introduction of an RNG to avail PET radionuclides for routine PET imaging would, of course, depend on regional and country regulations. Use of PET RNGs in radiopharmacies requires either dedicated radionuclide specific radio-synthesizers or kit-based, micro-scale radio-synthesizers that offer fully integrated radiochemistry solutions on the bench top to synthesize the radiotracers, and quality control (QC) testing equipment [e.g. gas chromatography (GC), analytical radio-high-performance liquid chromatography (radio-HPLC), dose calibrators, radio-thin layer chromatography (radio-TLC), etc.] to ensure the safety of the patient. Additionally, it also requires the availability specially trained and skilled personnel to operate the entire process, from generator elution to the aseptic dose preparation.

In light of the perceived need to address regulatory requirements and to achieve Current Good Manufacturing Practices (cGMP) compliance, the field of RNGs is increasingly migrating toward the use of automated systems. Automation offers several advantages, including providing robust, repeatable processes, reducing the radiation dose to operators, offering consistent performance of the generator system, ability to handle and distribute multiple doses daily, and providing a log of the steps performed in the elution of the medical radionuclide [217]. Automation also facilitates regulatory compliance through manufacturer design qualification/operational qualification/performance qualifications and scheduled maintenance protocols performed on automated RNGs by trained service engineers.

Due to the short half-life of most of the generator produced radionuclides, the prospect of using kit-type formulation strategies seemed sagacious, as it would reduce reagent preparation and setup time, curtail human errors, eliminate the possibility of contamination, facilitate cleanup, and simplify operation. A further advantage of kits is the simplification of reagent handling. Rather than individually managing multiple reagents, the radiopharmacy need only to manage the kit as a single unit, greatly simplifying compliance with FDA regulations concerning production of PET tracers for injection into humans. Kit formulation strategies are beginning to put the capability to make PET tracers directly into the hands of the scientists and clinicians who need them.

Summary and outlook

A review of the current use and potential applications of PET RNGs indicates that this dynamic field is evolving rapidly from being an "art rather than a science", to a radionuclide delivery system. This multidisciplinary technological paradigm, which resides at the intersection of radiochemistry and chemical separation techniques, has undergone a tremendous metamorphosis and is destined to bring perpetual changes in clinical PET practices. New PET generators are emerging persistently, taking shape progressively and after maturation, steadily migrating towards clinical practice. The relentless parade of new PET generators, destined to unfold on many new fronts in clinical PET, harbor boundless possibilities and a wellspring to spur myriad applications in clinical PET. Almost every advance in PET RNGs is billed as a breakthrough, and the list of "next generation generators" grows longer every day. While the potential benefits of the new PET generators are tremendous, so are the challenges of bringing them from laboratory research to clinical practice for their impact.

RNGs have their roots in NM, and progress is inextricably linked to the continued research and advancements of PET imaging in NM. As PET is moving to the forefront of NM, demands for new radionuclides are emerging far more quickly than they did over the past decade. Because of the pace at which the PET scene is evolving, RNG strategies need a vision for today and tomorrow. Given the importance of PET in molecular imaging, paradigm-changing PET RNGs are quick to grab the headlines.

The advances made so far are exciting and efforts to develop new generators for use in clinical practices are evolving persistently. While the clinical realization of any of these paradigm-changing new PET RNGs would be expected to revitalize current PET practices and provide myriad opportunities to instill new possibilities to reshape the clinical PET of today and tomorrow, it requires effective harnessing of emerging technology and inspired vision from clinical as well as industrial partners to provide functional systems in appropriate volumes at the right cost. In this premise, each new generator and associated daughter PET radionuclide must be evaluated on the basis of their performance, affordability, cost effectiveness, and propitious clinical outcome. "Synergy, understanding, and collaboration" have emerged as the three most common descriptors that underpin their survival and growth.

Much has been written about the ⁸²Sr/⁸²Rb generator due to its dominance in the field of cardiac PET, but this example should really be viewed largely as a model that can be copied for several alternative PET RNGs. They deserve greater attention not only because a greater range of PET radionuclides will be needed, but also for the adaptability to use different radionuclides for different applications. With a better understanding of the properties, including the physical and chemical properties of each of the radionuclides, a wide range of PET radiopharmaceuticals can be prepared which can be effectively used for molecular imaging of targets for a broad range of biological or disease processes. Despite being a late entrant to PET, the ⁶⁸Ge/⁶⁸Ga generator has not only consolidated its presence but also established a strong foothold at the forefront of PET. In a relatively short time-span, 68Ga has virtually pervaded all areas of PET and has become a key PET radionuclide of choice for the foreseeable future. It may be surmised that diffusion of ⁶⁸Ga tracers seem poised to bring spectacular developments and serve as the springboard to spur new breakthroughs in PET imaging.

The restricted clinical utility of PET in diagnostic NM is mainly due to limited availability of this technology, its cost, and limited published data supporting its use. In recent years, PET imaging has moved from an exotic diagnostic modality for very few patients to a mainstream modality. The future of generator derived PET radionuclides is, of course, difficult to predict, and there will be surprising inventions, as in the past, in which PET radionuclides availed from a generator may have an unexpected application that will continue to fuel development the field. Healthcare service providers should seize the opportunity by keeping their organizational strategies updated in the face of continually evolving PET radiotracer technologies and ensure that their organizations continue to pursue optimum PET radiotracers as well as innovative technologies in an effort to reduce the gap between requirements and capabilities.

In spite of the huge potential of PET RNGs, costeffective availability of parent radionuclides is one major challenge which currently obstructs the path for their widespread progress. To traverse these obstacles, a constant and reliable supply of parent radionuclides of the required quality in the desired quantities needs to be assured. Owing to the inherent requirement of capital intensive installation of cyclotrons, significant expertise, skilled manpower and adequate resources to undertake regular production of radionuclides, the number of commercial radioisotope suppliers remains finite and their current production capabilities remain limited. Additional research efforts and large resources are warranted to undertake their production on a regular basis to meet the current and foreseeable demands.

There are several reasons why marketing authorization of PET RNGs are not attractive. First, the economic investment costs for a PET RNG with a very small potential market cannot be recouped. Second, an RNG housing a shortlived parent radionuclide is unattractive and the investment required for its large-scale production would usually require the availability of targets as well as high energy accelerators. Furthermore, such short-lived parent radionuclides would have to be produced regularly to replenish the generators. This demands extremely efficient logistics for the overall process of availing the proven benefit of PET tracers.

The clinical success of a new RNG largely depends on its application, cost-effective availability of the parent radionuclide, the overall cost associated with its GMP production, and the process of obtaining regulatory approval, which is gradually increasing. The process of converting a promising RNG for the clinical application is rife with scientific and regulatory obstacles. In order to surmount the regulatory barrier and become accepted in daily clinical practice, PET RNGs will likely need a strong cor-

porate sponsor willing to invest. Industry sponsorship is possible in this era because academic inventors now routinely secure strong intellectual property rights on RNGs. If strong intellectual property rights are secured and industry makes a substantial investment, the cost of new PET RNGs will be significantly high, which may lead to few early users and consequently inhibit their widespread use as well as convergence into standard clinical practice. In order to mitigate such unforeseen situations, a more holistic, pragmatic and outcome-oriented regulatory approach is required. NM needs leaders who are capable of managing both technical and non-technical issues, whether in academia, business, government or the medical sector. It is the responsibility of all the stake holders, including research scientists, clinicians, radiopharmacists, hospitals and industries to share a common platform to convince regulatory authorities of the need for change to address the challenges in an ever-demanding regulatory environment. While the utility of generator derived PET tracers has passed many milestones and made considerable inroads into the arena of PET imaging, but without a doubt this is just the tip of the iceberg and further excitement in this field is promised in the foreseeable future.

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Disclosure of conflict of interest

None.

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